

[³H]gallopil as above in a 1-mL total assay volume. The remaining procedures were as described above. Each data point shown represents the mean of triplicate determinations. Each experiment was repeated at least 3 times.

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Registry No. (±)-1-HCl, 23313-68-0; (±)-2, 102852-52-8; (±)-3, 102852-53-9; (±)3-HCl, 102852-54-0; Ca, 7440-70-2.

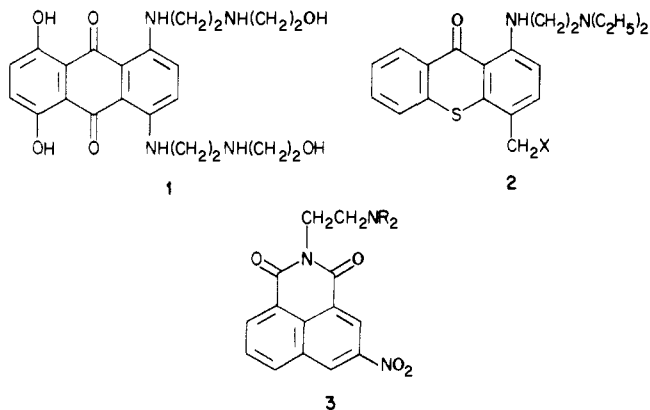
Amino-Substituted *p*-Benzoquinones

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Based on the observation of outstanding antineoplastic activity of a number of amino-substituted anthraquinones, thioxanthones, and *N*-(aminoethyl)-substituted naphthalimides, four types of amino-substituted *p*-benzoquinones were designed, synthesized, and their biological activity evaluated. Although none of these compounds exhibited inhibitory activity against P388 leukemia, 2,5-bis[[4-[(dimethylamino)methyl]phenyl]amino]-3,6-dibromo-1,4-benzoquinone and the corresponding dichloro compound demonstrated good inhibitory activity against the proliferating human colon adenocarcinoma in vitro. The dichloro compound was also found to be active against the leukemia L1210 screening in vitro. 2,5-Bis[[2-(dimethylamino)ethyl]amino]-1,4-benzoquinone possessed inhibitory activity against *Neisseria catarrhali*.

A common *o*-aminoquinoid unit was reported¹ among several antitumor antibiotics including streptonigrin, actinomycin D, the mitomycins, and porfiromycin. On the basis of this concept, the AB ring units of streptonigrin^{2,3} and amino-containing benzoquinones⁴ and naphthoquinones⁵ were synthesized and evaluated for their biological activity. Knowledge gained through these studies led to the proposition of a N-O-O working hypothesis.⁶ Adamson⁷ subsequently postulated a structural modification approach for the anthracyclines including adriamycin and daunomycin. Repeated structural designs and syntheses, together with a structural lead from the National Cancer Institute,⁸ culminated in the synthesis of a dihydroxylated amino-containing anthraquinone DHAQ,⁹ 1,4-dihydroxy-5,8-bis[[2-[(2-hydroxyethyl)amino]ethyl]amino]anthraquinone (1), which displayed outstanding anticancer activity and is being evaluated in cancer patients.¹⁰⁻¹²



The DHAQ side chain contains an (aminoethyl)amino unit in common with two other types of antineoplastic agents 2 (e.g., hycanthone,^{13,14} X = OH) and 3 (e.g., mitonafide,¹⁵⁻¹⁷ R = CH₃). The presence of the distal nitrogen atom among the three structurally different ring systems,⁸⁻¹⁴ together with the report that DNA binding or

intercalation may not be the true mechanism of action for the anticancer activity of DHAQ,¹⁸ suggested that incorporation of the substituted (aminoethyl)amino group and related side chains into other simpler ring systems should be studied. Consequently, synthesis of compounds containing the following amino-substituted side chains attached to the opposite sides of *p*-benzoquinone was conducted.

Type I (Compounds 4a-c). *p*-Benzoquinones containing the 2-[(dimethylamino)ethyl]amino side chains with or without substituents at other positions.

Type II (Compounds 5a,b). The ethylene linkage of

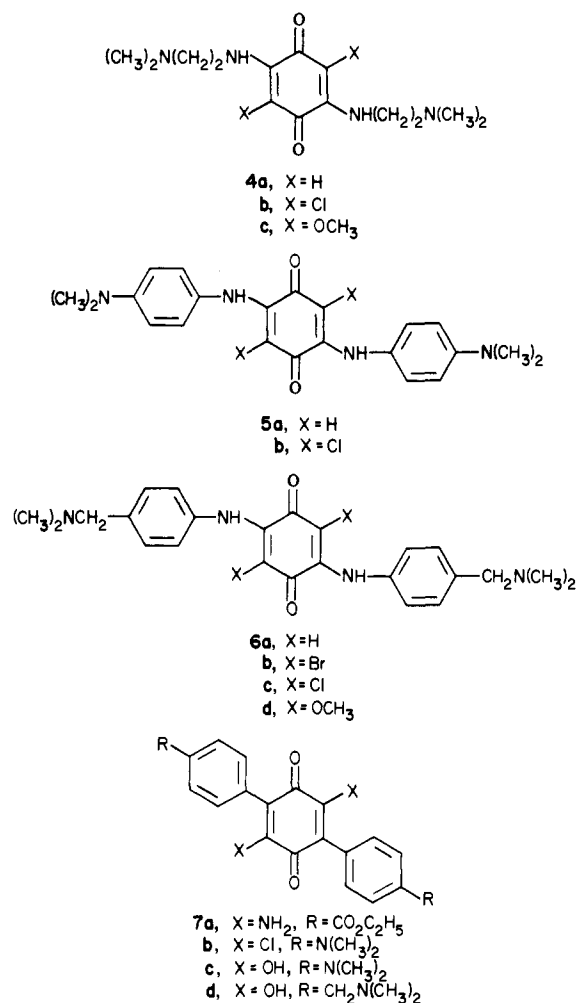
- (1) Rao, K. V.; Biemann, K.; Woodward, R. B. *J. Am. Chem. Soc.* **1963**, *85*, 2532.
- (2) Liao, T. K.; Nyberg, W. H.; Cheng, C. C. *Angew. Chem.* **1967**, *79*, 100.
- (3) Liao, T. K.; Nyberg, W. H.; Cheng, C. C. *J. Heterocycl. Chem.* **1976**, *13*, 1063.
- (4) Zee-Cheng, K.-Y.; Cheng, C. C. *J. Med. Chem.* **1970**, *13*, 264.
- (5) Podrebarac, E. G.; Cheng, C. C. *J. Org. Chem.* **1970**, *35*, 281.
- (6) Zee-Cheng, K.-Y.; Cheng, C. C. *J. Pharm. Sci.* **1970**, *59*, 1630.
- (7) Adamson, R. H. *Cancer Chemother. Rep.* **1974**, *58*, 293.
- (8) Cheng, C. C.; Zee-Cheng, R. K.-Y.; Narayanan, V. L.; Ing, R. B.; Paull, K. D. *Trends Pharmacol. Sci.* **1981**, *2*, 223.
- (9) Zee-Cheng, R. K.-Y.; Cheng, C. C. *J. Med. Chem.* **1978**, *21*, 291.
- (10) Traganos, F. *Pharmacol. Ther.* **1983**, *22*, 199.
- (11) Smith, I. E. *Cancer Treat. Rev.* **1983**, *10*, 103.
- (12) Cheng, C. C.; Zee-Cheng, R. K.-Y. *Prog. Med. Chem.* **1983**, *20*, 83.
- (13) Rosi, D.; Peruzzotti, G.; Dennis, E. W.; Berberian, D. A.; Freele, H.; Archer, S. *Nature (London)* **1965**, *208*, 1005.
- (14) Archer, S.; Zayed, A.-H.; Rej, R.; Rugino, T. A. *J. Med. Chem.* **1983**, *26*, 1240.
- (15) Braña, M. F.; Castellano, J. M.; Jiménez, A.; Llombart, A.; Rabadán, F. P.; Roldán, M.; Roldán, C.; Santos, A.; Vázquez, D. *Curr. Chemother. Proc. Int. Congr. Chemother. 10th* **1977**, *2*, 1216.
- (16) Idoipe, A.; Santillán, M. S.; Martínez-Larrañaga, M. R.; González, E.; de Jalón, P. D. G. *Arch. Farmacol. Toxicol.* **1979**, *17*, 8.
- (17) Braña, M. F.; Castellano, J. M.; Roldán, C. M.; Santos, A.; Vázquez, D.; Jiménez, A. *Cancer Chemother. Pharmacol.* **1980**, *4*, 61.
- (18) Johnson, R. K.; Zee-Cheng, R. K.-Y.; Lee, W. W.; Acton, E. M.; Henry, D. W.; Cheng, C. C. *Cancer Treat. Rep.* **1979**, *63*, 425.

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type I compounds is replaced by a *p*-phenylene unit, i.e., compounds containing the *p*-(dimethylamino)anilino functions.

Type III (Compounds 6a-d). To increase the basicity of the distal amino function of compounds of type II, a methylene linkage is inserted between the distal amino and the phenyl linkage, i.e., compounds containing the *p*-[(dimethylamino)methyl]anilino functions.

In addition, several amino-containing *p*-benzoquinone derivatives having bis(substituted-phenyl) functions attached to the opposite positions (7) were also prepared. These compounds are analogues of polyporic acid that have shown inhibitory activity against Walker 256 carcinosarcoma in rats.⁴



Chemistry. Compounds 4a, 5a, and 6a were prepared by treating the appropriate amine with excess 1,4-benzoquinone. Excess quinone rather than the amine was used because the reaction of this type involves the 1,4-addition of the amine to the quinone, followed by rearrangement to the substituted hydroquinone and oxidation by a second molecule of quinone,^{19,20} hence, the unsubstituted 1,4-hydroquinone is produced as a byproduct. The use of excess quinone could be avoided by passing O₂ into the reaction mixture.²¹ Depending on the availability of the appropriate quinones or the amines, reaction conditions could be selected based on the aforementioned information; for better results, the reaction temperature should

be kept at room temperature or below. Compounds 4b, 5b, and 6c were obtained from chloranil and 2 equiv of the amine. In general, for the aliphatic amines, room temperature and the use of nonpolar solvents are preferred, but more polar solvents, higher reaction temperature, and excess amines are desirable for the aromatic amines. The corresponding dibromo derivative 6b was prepared in a similar manner. The dimethoxy derivatives 4c and 6d were prepared from 2,3,5,6-tetramethoxy-1,4-benzoquinone and excess amine at elevated reaction temperature.

Treatment of 2,5-dichloro-1,4-benzoquinone with the diazonium salt of ethyl *p*-aminobenzoate yielded 2,5-bis[4-(ethoxycarbonyl)phenyl]-3,6-dichloro-1,4-benzoquinone, which was converted to the corresponding diamino derivative 7a by using ethanolic NH₃. Compound 7b was prepared from the diazonium salt of 4-(dimethylamino)aniline. In this experiment, the pH of the reaction mixture is very critical. Optimum pH was found to be 4.7, since at higher pH a polymeric material resulted and at lower pH the reaction proceeded very slowly and led to the formation of undesired compounds. The pH of the reaction mixture, therefore, should be checked occasionally and adjusted to 4.7 with NaOAc solution.

Base hydrolysis of compound 7b readily yielded 7c, the polyporic acid⁴ analogue. Compound 7d was obtained by the base hydrolysis of 2,5-bis(dimethylamino)-3,6-bis[4-[(dimethylamino)methyl]phenyl]-1,4-benzoquinone. The latter was prepared by the bromination of 2,5-dichloro-3,6-bis(4-tolyl)-1,4-benzoquinone⁴ with *N*-bromosuccinimide followed by the treatment with dimethylamine.

Biological Activity. Among the benzoquinones synthesized, 2,5-bis[4-[(dimethylamino)methyl]phenyl]amino]-3,6-dibromo-1,4-benzoquinone (6b) showed significant inhibitory activity against human colon adenocarcinoma in vitro (proliferating) at an average ID₅₀ of 1.4 × 10⁻⁷ M. The corresponding dichloro compound 6c had an average ID₅₀ value of 1.0 × 10⁻⁷ M. In addition, 6c demonstrated outstanding inhibitory activity against the leukemia L1210 in vitro²² with an average ID₅₀ value of 3.7 × 10⁻⁸ M, which is comparable to the activity of mithramycin (ID₅₀ = 8.0 × 10⁻⁸ M). Borderline activity against the human colon adenocarcinoma was noted with compound 6a and against in vitro L1210 with compound 6d. In the antibacterial/antifungal screen, 2,5-bis[[2-(dimethylamino)ethyl]amino]-1,4-benzoquinone (4a) was found to have inhibitory activity against *Neisseria catarrhali*. None of the benzoquinones possess inhibitory activity against the NCI leukemia L1210 screening.

Experimental Section

All melting points were taken on a Thomas-Hoover melting point apparatus. Where analyses are indicated only by symbols of the elements, analytical results obtained for these elements were within ±0.4% of the theoretical values.

2,5-Bis[[2-(dimethylamino)ethyl]amino]-1,4-benzoquinone (4a). To a stirred solution of 10.8 g (100 mmol) of 1,4-benzoquinone in 300 mL of EtOH was added dropwise 4.4 g (50 mmol) of [2-(dimethylamino)ethyl]amine in 10 mL of EtOH. The mixture was stirred at room temperature for 7 h and allowed to stand overnight. It was then concentrated to 100 mL. To the solution was added 100 mL of Et₂O. A small amount of polar impurity deposited was removed by filtration. The filtrate was diluted with 200 mL of Et₂O and evaporated slowly under reduced pressure. The solid that formed was collected by filtration and dried to give 9.1 g (59% yield) of crude 4a, mp 138–141 °C. It was recrystallized from a mixture of EtOH and petroleum ether (bp 35–40 °C) to give 6.2 g (44% yield) of 4a: mp 157–159 °C; UV λ_{max} (MeOH) 210 nm (log ε 4.27), 245 (3.51), 340 (4.22); NMR

(19) Heller, G. *Ann. Chem.* 1919, 418, 259.

(20) Cason, J. *Org. Reactions* 1948, IV, 305.

(21) Cavallito, C. J.; Soria, A. E.; Hoppe, J. O. *J. Am. Chem. Soc.* 1950, 72, 2661.

(22) Baguley, B. C.; Nash, R. *Eur. J. Cancer* 1981, 17, 671.

(CDCl₃) δ 2.2 (s, 12 H, CH₃), 2.8 (t, 4 H, CH₂), 3.2 (t, 4 H, CH₂), 5.25 (s, 2 H, vinyl H), 6.9 (br s, 2 H, NH). Anal. (C₁₄H₂₄N₄O₂) C, H, N.

2,5-Bis[[2-(dimethylamino)ethyl]amino]-3,6-dichloro-1,4-benzoquinone Dihydrochloride (4b). To a stirred solution of 5 g (20 mmol) of chloranil in 100 mL of C₆H₆ was added dropwise at room temperature a solution of 3.6 g (40 mmol) of [2-(dimethylamino)ethyl]amine in 50 mL of C₆H₆ in 1 h. The color of the reaction mixture gradually changed to dark brown. Stirring was continued overnight. The resulting brown solid was collected by filtration and washed with Et₂O (2 × 20 mL) to give 7 g (95% yield). An analytical sample was obtained by recrystallization from MeOH: mp 250–252 °C; UV λ_{\max} (MeOH) 226 nm (log ϵ 4.30), 354 (4.39); NMR (CDCl₃) δ 2.3 (s, 12 H, NCH₃), 3.5 (q, 2 H, CH₂), 3.9 (q, 2 H, CH₂), 7.6 (br s, 2 H, NH). Anal. (C₁₄H₂₂Cl₂N₄O₂·2HCl) C, H, N.

2,5-Bis[[2-(dimethylamino)ethyl]amino]-3,6-dimethoxy-1,4-benzoquinone Dihydrochloride (4c). To a warm (45–50 °C) solution of 1.2 g (5 mmol) of 2,3,5,6-tetramethoxy-1,4-benzoquinone in a mixture of 40 mL of MeOH and 2 mL of H₂O was added, with stirring, a solution of 1.2 g (14 mmol) of [2-(dimethylamino)ethyl]amine in 5 mL of MeOH. The mixture was refluxed for 5 min, then allowed to stir at room temperature overnight. The resulting solution was evaporated to dryness under reduced pressure. The residue was triturated with 50 mL of petroleum ether (bp 35–40 °C) and the solid collected by filtration to give 1.7 g (95% yield) of the free base of 4c as a dark-purple solid, mp 134–136 °C. One-half gram of the product was dissolved in a mixture of 40 mL of MeOH and 30 mL of petroleum ether (bp 35–40 °C). To this was added 2 mL of HCl in MeOH (3.5 mmol of HCl/mL of MeOH). The mixture was diluted with 20 mL of Et₂O. The precipitated solid was collected by filtration and washed with Et₂O (2 × 10 mL) to give 0.52 g of 4c, mp 230–232 °C. Recrystallization from a mixture of MeOH and Et₂O yielded 0.4 g of analytically pure product, mp 232–234 °C. Compound 4c was found to be readily soluble in H₂O or MeOH: UV λ_{\max} (MeOH) 228 nm (log ϵ 4.39), 360 (4.34). Anal. (C₁₈H₂₈N₄O₄·2HCl·H₂O) C, H, N.

2,5-Bis[[4-(dimethylamino)phenyl]amino]-1,4-benzoquinone (5a). This product was reported previously¹⁹ without much detailed information on procedure, product characterization, and analyses. To a stirred solution of 9.7 g (90 mmol) of 1,4-benzoquinone in 100 mL of EtOH was added 10.5 g (50 mmol) of 4-(dimethylamino)aniline hydrochloride followed by 12.5 g of AcONa. The mixture was stirred at room temperature for 2 days. The resulting precipitate was collected by filtration, washed with EtOH and H₂O, and dried to give 9.6 g (51% yield) of 5a, mp >300 °C. It was recrystallized from AcOH to afford 6 g of analytically pure 5a: mp >300 °C; UV λ_{\max} (CHCl₃) 270 nm (log ϵ 4.49), 318 (4.40) 485 (4.15); mass spectrum, *m/e* 376 (M⁺). Anal. (C₂₂H₂₄N₄O₂·0.25H₂O) C, H, N.

2,5-Bis[[4-(dimethylamino)phenyl]amino]-3,6-dichloro-1,4-benzoquinone (5b). To a stirred solution of 6.1 g (25 mmol) of chloranil in 350 mL of EtOH was added 8.4 g (62 mmol) of 4-(dimethylamino)aniline. The mixture was refluxed for 90 min, then was allowed to stir at room temperature overnight. The reaction mixture was concentrated to one-third of its original volume, and the resulting precipitate was collected by filtration. It was then washed with Et₂O and dried to give 10.1 g of light-green product. The solid was suspended in NH₄OH, stirred for 30 min, and filtered. It was washed with H₂O and dried to give 7.7 g (70% yield) of 5b: mp >300 °C; UV λ_{\max} (CHCl₃) 270 nm (log ϵ 4.50), 340 (4.27), 530 (4.03). Anal. (C₂₂H₂₂Cl₂N₄O₂) C, H, N.

2,5-Bis[[4-[(dimethylamino)methyl]phenyl]amino]-1,4-benzoquinone (6a). A mixture of 5.61 g (37 mmol) of 1,4-benzoquinone, 3.8 g (18 mmol) of 4-[(dimethylamino)methyl]aniline, and 200 mL of EtOH was refluxed for 5.5 h, then allowed to stir overnight. The solid product was collected by filtration (1.69 g). The mother liquor was evaporated under reduced pressure, and the residue was triturated with 100 mL of EtOH to yield another 0.65 g of 6a (total yield 2.34 g, 31%), mp >300 °C. The product was recrystallized from a mixture of EtOH and CHCl₃: UV λ_{\max} (MeOH) 273 nm (log ϵ 4.26), 388 (4.28); NMR (CDCl₃) δ 2.30 (12 H, N-CH₃), 3.40 (s, 4 H, CH₂), 6.10 (2 H, vinyl-H), 7.10–7.50 (m, 8 H, Ar-H), 8.15 (2 H, NH); mass spectrum,

m/e 404 (M⁺). The free base 6a was dissolved in CHCl₃, and methanolic HCl was added until pH 1 was reached. The solid formed was collected by filtration and purified by recrystallization from MeOH to give the dihydrochloride salt of 6a: mp >330 °C. Anal. (C₂₄H₂₈N₄O₂·2HCl·0.5H₂O) C, H, N.

2,5-Bis[[4-[(dimethylamino)methyl]phenyl]amino]-3,6-dibromo-1,4-benzoquinone (6b). To a stirred and boiling solution of 5.9 g (14 mmol) of 2,3,5,6-tetrabromo-1,4-benzoquinone in 500 mL of EtOH was added 4.5 g (30 mmol) of 4-[(dimethylamino)methyl]aniline. The resulting mixture was refluxed for 45 min and allowed to cool overnight. The solid product was collected by filtration, washed with EtOH, and dried to give 8.1 g of the hydrobromide salt of 6b, mp >300 °C. It was dissolved in 500 mL of H₂O. A small amount of insoluble material was separated by filtration. The filtrate was basified with 10% NaHCO₃ and extracted with CHCl₃ (4 × 100 mL). After removal of the solvent, 4.5 g (57% yield) of 6b was obtained: mp >300 °C; UV λ_{\max} (MeOH) 252 nm (log ϵ 4.26), 395 (4.17); NMR (CDCl₃) δ 2.3 (s, 12 H, CH₃), 3.4 (s, 4 H, CH₂), 7.0–7.5 (m, 8 H, Ar-H), 8.2–8.8 (br s, 2 H, NH); mass spectrum *m/e* 560 (M⁺), 561, 563. Anal. (C₂₄H₂₆Br₂N₄O₂·H₂O) C, H, N.

The free base (2.8 g) was dissolved in 250 mL of CHCl₃. To the solution was added methanolic HCl until pH 1 was reached. The resulting precipitate was collected by filtration, washed with CHCl₃ and EtOH, and dried to give 3.4 g of the dihydrochloride salt of 6b: mp >3000 °C. Anal. (C₂₄H₂₆Br₂N₄O₂·2HCl·1.5H₂O) C, H, N.

2,5-Bis[[4-[(dimethylamino)methyl]phenyl]amino]-3,6-dichloro-1,4-benzoquinone (6c). To a boiling solution of 2.0 g (8.1 mmol) of chloranil in 150 mL of EtOH was added, with stirring, 2.65 g (18 mmol) of 4-[(dimethylamino)methyl]aniline. The resulting solution was refluxed for 40 min and chilled overnight. The solid product was collected by filtration, washed with EtOH, and dried to yield 4.4 g of the dihydrochloride salt of 6c, mp >300 °C. The salt was dissolved in 1% dilute HCl solution; a minute amount of insoluble material was removed by filtration, and the filtrate was made basic with NaHCO₃. It was extracted immediately with CHCl₃ (3 × 50 mL). After drying over anhydrous Na₂SO₄ and removing solvent, the residue was recrystallized from a mixture of CHCl₃ and EtOH to give 2.75 g (69% yield) of 6c: mp >300 °C; UV λ_{\max} (MeOH) 270 nm (log ϵ 4.32), 395 (4.23); NMR (CDCl₃) δ 2.3 (s, 12 H, NCH₃), 3.5 (s, 4 H, CH₂), 7.2–7.6 (m, 8 H, Ar-H); mass spectrum *m/e* 472 (M⁺ - 1), 474 (M⁺ + 1). Anal. (C₂₄H₂₆Cl₂N₄O₂·H₂O) C, H, N.

2,5-Bis[[4-[(dimethylamino)methyl]phenyl]amino]-3,6-dimethoxy-1,4-benzoquinone (6d). A mixture of 3.41 g (15 mmol) of 2,3,5,6-tetramethoxy-1,4-benzoquinone and 5.50 g (37 mmol) of 4-[(dimethylamino)methyl]aniline in 275 mL of MeOH was refluxed for 24 h. Solvent was removed and the residue triturated with Et₂O. The resulting solid was collected by filtration to give 3.25 g of dark-green product 6d. It was dissolved in 200 mL of CHCl₃, and the volume of the solution was reduced to 30 mL. To the solution was added 50 mL of petroleum ether (bp 35–40 °C). A small amount of solid was removed by filtration. To the filtrate was added an additional 100 mL of petroleum ether, and the mixture was allowed to stand overnight. The resulting solid was collected by filtration and dried to give 2.6 g (55% yield) of 6d: mp 165–168 °C; UV λ_{\max} (MeOH) 276 nm (log ϵ 4.51), 408 (4.24); NMR (CDCl₃) δ 2.3 (s, 12 H, CH₃), 3.3 (s, 6 H, OCH₃), 3.5 (s, 4 H, CH₂), 7.0–7.4 (m, 8 H, Ar-H), 7.7 (br s, 2 H, NH). Anal. (C₂₆H₃₂N₄O₄) C, H, N.

2,5-Bis[4-(ethoxycarbonyl)phenyl]-3,6-dichloro-1,4-benzoquinone. The diazonium salt of ethyl *p*-aminobenzoate was prepared by adding a solution of 13.2 g of NaNO₂ in 15 mL of H₂O to a cooled solution (-5 to 0 °C) of 29.7 g (180 mmol) of the amine in 36 mL of HCl and 30 mL of H₂O. The diazonium chloride thus prepared was added dropwise, with stirring, to 10.5 g (60 mmol) of 2,5-dichloro-1,4-benzoquinone in a mixture of 400 mL of MeOH and 130 mL of Et₂O at 5–10 °C. During the addition, a solution of 36 g (440 mmol) of NaOAc in 100 mL of H₂O was simultaneously added dropwise so that the reaction mixture remained neutral throughout the addition. After addition, the mixture was stirred at room temperature for 20 h, concentrated under reduced pressure to ca. 150 mL, and poured into 1.5 L of H₂O. The resulting solid was collected by filtration, washed with H₂O, dried, and then washed with petroleum ether (bp 35–40 °C)

to give 28 g of crude product. It was recrystallized from a mixture of CHCl_3 and petroleum ether to give 9.8 g (35% yield) of the chloroester: mp 248–250 °C; NMR ($\text{CF}_3\text{CO}_2\text{H}$) δ 1.5 (t, 6 H, CH_3), 4.6 (q, 4 H, CH_2), 7.8 (ABq, 8 H, $J = 7$ cps, Ar-H); mass spectrum m/e 472 ($M^+ - 1$), 474 ($M^+ + 1$). Anal. ($\text{C}_{24}\text{H}_{18}\text{Cl}_2\text{O}_6 \cdot 0.5\text{H}_2\text{O}$) C, H.

2,5-Bis[4-(ethoxycarbonyl)phenyl]-3,6-diamino-1,4-benzoquinone (7a). A solution of 5 g (10 mmol) of the aforementioned chloroester in 150 mL of 18% ethanolic NH_3 and 175 mL of EtOH was stirred at 0 °C for 2 h, then at room temperature for 30 h. The resulting solid was collected by filtration, washed with H_2O , and dried to give 4.0 g of crude **7a**, mp 294–297 °C. Recrystallization from 150 mL of $\text{HCON}(\text{CH}_3)_2$ afforded 3.5 g (83% yield) of **7a** mp 303–305 °C; UV λ_{max} (MeOH) 228 nm ($\log \epsilon$ 4.48), 335 (4.44). Anal. ($\text{C}_{24}\text{H}_{22}\text{N}_2\text{O}_6$) C, H, N.

2,5-Bis[4-(dimethylamino)phenyl]-3,6-dichloro-1,4-benzoquinone (7b). To a solution of 23 g (110 mmol) of 4-(dimethylamino)aniline hydrochloride in 10 mL of H_2O and 15 mL of concentrated HCl cooled to –5 to 0 °C was added dropwise, in 3 h, a solution of 7.8 g (110 mmol) of NaNO_2 in 20 mL of H_2O . The diazonium chloride thus prepared was added simultaneously with 15 g (180 mmol) of NaOAc in 35 mL of H_2O to a solution of 5.5 g (31 mmol) of 2,5-dichloro-1,4-benzoquinone in 200 mL of MeOH and 120 mL of Et₂O at 12–15 °C. The pH of the reaction mixture was kept at 4.6. After the addition, the pH of the mixture was carefully adjusted to 4.7, and the reaction mixture was stirred at room temperature for 16 h, then at 60–65 °C for 5 h. It was concentrated under reduced pressure to almost dryness, then added into 1.5 L of 1% aqueous NaOAc. The resulting solid was collected by filtration, washed with H_2O , and dried to give 11.4 g of crude product. It was recrystallized from 250 mL of CHCl_3 to give 0.96 g of crystallized **7b**. The mother liquor was concentrated and chromatographed over SiO_2 column (22 cm, 100 g) and eluted successively with 1:1 C_6H_6 – CHCl_3 and CHCl_3 . From the CHCl_3 fraction, an additional 1.60 g of the same product was collected. The total yield of **7b** was therefore 20%: mp >300 °C; UV λ_{max} (MeOH) 265 nm ($\log \epsilon$ 4.18), 295 (4.22), 570 (3.61); mass spectrum, m/e 414 ($M^+ - 1$), 416 ($M^+ + 1$). Anal. ($\text{C}_{22}\text{H}_{20}\text{Cl}_2\text{N}_2 \cdot 1.5\text{H}_2\text{O}$) C, H, N.

From the initial C_6H_6 – CHCl_3 elution there was obtained 0.7 g of the mono-aryl-substituted 2,4-dichloro-3-(4-dimethylamino)phenyl-1,4-benzoquinone: mp 188–190 °C; NMR (CDCl_3) δ 3.1 (s, 6 H, NCH_3), 6.7–7.5 (m, 5 H, Ar- & vinyl-H). Anal. ($\text{C}_{14}\text{H}_{11}\text{Cl}_2\text{NO}_2$) C, H, N.

2,5-Bis[4-(dimethylamino)phenyl]-3,6-dihydroxy-1,4-benzoquinone (7c). To a hot (55 °C) solution of 200 mL of 10% NaOH was added, with stirring, a slurry of 2.2 g (5 mmol) of **7b** in 150 mL of MeOH. The mixture was stirred and refluxed for 75 min and filtered while hot. The filtrate was cooled and acidified with 10% HCl to pH 2, then made basic (to pH 8) with saturated NaOAc solution. The resulting solid was collected by filtration, washed with a small amount of H_2O , and dried. The product was purified by recrystallization from CHCl_3 , mp >300 °C. The yield was 1.38 g (69%): UV λ_{max} (MeOH) 292 nm ($\log \epsilon$ 4.55), 460 (3.35); mass spectrum, m/e 378 (M^+). Anal. ($\text{C}_{22}\text{H}_{22}\text{N}_2\text{O}_4$) C, H, N.

2,5-Bis(dimethylamino)-3,6-bis[4-[(dimethylamino)methyl]phenyl]-1,4-benzoquinone. A mixture of 6 g (17 mmol) of 2,5-dichloro-3,6-bis(4-tolyl)-1,4-benzoquinone⁴ and 6 g (34 mmol) of *N*-bromosuccinimide in 650 mL of CCl_4 containing 200 mg of benzoyl peroxide was refluxed for 16 h. Solvent was removed

under reduced pressure, and the residue was dissolved in 200 mL of CH_3CN . The solution was cooled to 0–5 °C, and dry Me_2NH was bubbled into the mixture for 90 min (ca. 1–2 bubbles/s). The reaction mixture was stirred at room temperature for 24 h and evaporated under reduced pressure. The resulting viscous residue was dissolved in 200 mL of Et₂O and extracted with 10% HCl (5 × 50 mL). The acid extract was made basic with saturated NaHCO_3 solution, then extracted with CHCl_3 (3 × 100 mL). The CHCl_3 extract was dried (Na_2SO_4), and solvent was removed to yield 5.0 g of the crude product (mp 187–190 °C), which was recrystallized from a mixture of EtOH and CHCl_3 to give 3.7 g (48% yield) of pure product: mp 207–209 °C dec; UV λ_{max} (MeOH) 305 nm ($\log \epsilon$ 4.10), 370 (4.24); NMR (CDCl_3) δ 2.20 (br s, 12 H, benzylic NCH_3), 2.85 (br s, 12 H, NCH_3), 3.40 (br s, 4 H, CH_2), 7.0–7.8 (m, 8 H, Ar-H). The product was used in the following experiment.

2,5-Bis[4-[(dimethylamino)methyl]phenyl]-3,6-dihydroxy-1,4-benzoquinone (7d). A mixture of 2.3 g (5 mmol) of the preceding product and 80 mL of 25% NaOH was refluxed for 2.5 h. The mixture was cooled, and the solid was collected by filtration. It was dissolved in 30 mL of H_2O , and the insoluble material was removed by filtration. The filtrate was carefully neutralized to pH 7.2 with dilute HCl, and the precipitate formed was collected by filtration and dried to give 1.34 g (66% yield) of **7d**: mp >300 °C; UV λ_{max} (MeOH) 308 nm ($\log \epsilon$ 3.32), 530 (3.18). The dihydrochloride salt of **7d** was prepared by dissolving **7d** in CHCl_3 , and to the solution was added methanolic HCl to pH 1. The mixture was allowed to stand overnight, and the precipitated solid was collected by filtration. It was recrystallized from a mixture of MeOH and CHCl_3 to give 1.32 g of the dihydrochloride salt of **7d**: mp >300 °C. Anal. ($\text{C}_{24}\text{H}_{26}\text{N}_2\text{O}_4 \cdot 2\text{HCl} \cdot 1.25\text{H}_2\text{O}$) C, H, N.

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Registry No. **4a**, 102780-56-3; **4b**, 102780-57-4; **4c**, 102780-59-6; **4c**·2HCl, 102780-58-5; **5a**, 102780-60-9; **5b**, 95887-85-7; **6a**, 102780-61-0; **6a**·2HCl, 102780-62-1; **6b**, 102780-63-2; **6b**·HBr, 102780-64-3; **6b**·2HCl, 102780-65-4; **6c**, 102780-66-5; **6c**·2HCl, 102780-67-6; **6d**, 102780-68-7; **7a**, 102780-70-1; **7b**, 102780-71-2; **7b** (monosubstituted), 102780-72-3; **7c**, 102780-73-4; **7d**, 102780-75-6; **7d**·2HCl, 102780-76-7; $\text{H}_2\text{N}(\text{CH}_2)_2\text{NMe}_2$, 108-00-9; 4- $\text{H}_2\text{NC}_6\text{H}_4\text{NMe}_2$, 99-98-9; 4- $\text{H}_2\text{NC}_6\text{H}_4\text{NMe}_2\text{HCl}$, 2052-46-2; 4- $\text{H}_2\text{NC}_6\text{H}_4\text{CH}_2\text{NMe}_2$, 6406-74-2; 4- $\text{MeCH}_2\text{O}_2\text{CC}_6\text{H}_4\text{N}_2^+\text{Cl}^-$, 2028-79-7; 4- $\text{Me}_2\text{NC}_6\text{H}_4\text{N}_2^+\text{Cl}^-$, 100-04-9; 1,4-benzoquinone, 106-51-4; chloranil, 118-75-2; 2,3,5,6-tetrabromo-1,4-benzoquinone, 488-48-2; 2,5-bis[4-(ethoxycarbonyl)phenyl]-3,6-dichloro-1,4-benzoquinone, 102780-69-8; 2,5-dichloro-1,4-benzoquinone, 615-93-0; 2,5-bis[4-(dimethylamino)-3,6-bis[4-[(dimethylamino)methyl]phenyl]-1,4-benzoquinone, 102780-74-5; 2,5-dichloro-3,6-bis(4-tolyl)-1,4-benzoquinone, 28293-34-7.